

## **ACROSS**

	A harry to an explain to an term
1	A base is any molecule or ion
	that can form a new coordinate covalent
	bond, by donating a pair of electrons.
5	A is a molecule or ensemble
	formed by the combination of ligands and
	metal ions.
7	A is an atom, ion, or molecule
	donating one or more of its electrons
	through a coordinate covalent bond to one
	or more central atoms or ions
9	is the binding or complexation
	of a bi- or multidentate ligand.
10	chemistry is the study of
	chemical compounds containing bonds
	hetween carbon and a metal

## **DOWN**

2	The field of chemistry covers all
	chemical compounds except the myriad
	organic compounds, which are the
	subjects of organic chemistry.
3	A covalent bond is a type of
	covalent bonding between two atoms in
	which both electrons shared in the bond
	come from the same atom.
4	number is defined as the total
	number of neighbors of a central atom in a
	chemical compound.
6	A ligand is an atom or a
	polyatomic entity that connects two or
	more metal centres in a complex.
8	Arising from the size of the valance shell
	of a transition metal, the 18 rule
	is a rule of thumb used primarily in

transition metal chemistry for

of metal complexes.

characterizing and predicting the stability